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Biology Brooker 2nd Edition Study

Covalent Bonds. In a covalent bond, the atoms are bound by shared electrons. In a true covalent bond, the electronegativity values are the same (e.g., H 2, O 3), although in practice the electronegativity values just need to be close. If the electron is shared equally between the atoms forming a covalent bond, then the bond is said to be nonpolar.. Usually, an electron is more attracted to one ...

Ionic vs Covalent Bonds - Understand the Difference

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One Canberra study found that 98.5% of the rabbits taken were adults. In the largest study near Canberra, over 5.5 years, 19.3% of the diet of wedge-tailed eagles was comprised by rabbits (12.7% of prey biomass) among 1421 prey items, so the eagles took a total of some 275 rabbits in the 11 to 17 studied territories of the area.

Wedge-tailed eagle - Wikipedia

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